

JEE ADVANCED-2016 PAPER-2 PART - I: PHYSICS

1. The electrostatic energy of Z protons uniformly distributed throughout a spherical nucleus of radius R is given by $E = \frac{3}{5} \frac{Z(Z-1)e^2}{4\pi\epsilon_c R}$.

The measured masses of the neutron, ${}^{1}_{1}H$, ${}^{15}_{7}N$ and ${}^{15}_{8}O$ are 1.008665 u, 1.007825 u, 15.000109 u and 15.003065u, respectively. Given that the radii of both the $^{15}_{7}N$ and $^{15}_{8}O$ nuclei are same. 1u = 931.5 MeV/c² (c is the speed of light) and $e^2/(4\pi\epsilon_0) = 1.44$ MeV fm. Assuming that the difference between the binding energies of $^{15}_{7}\mathrm{N}$ and $^{15}_{8}\mathrm{O}$ is purely due to the electrostatic energy, the radius of either of the nuclei is. $(1fm=10^{-15}m)$

Sol. (C)

$$E_0 = \frac{3}{5} \times \frac{8 \times 7}{R} \times \frac{e^2}{4\pi\epsilon_0} = \frac{3}{5} \times \frac{8 \times 7}{R} \times 1.44 \text{MeV}$$

$$E_{N} = \frac{3}{5} \times \frac{7 \times 6}{R} \times \frac{e^{2}}{4\pi\epsilon_{0}} = \frac{3}{5} \times \frac{7 \times 6}{R} \times 1.44 MeV$$

so
$$|E_0 - E_N| = \frac{3}{5} \times \frac{1.44}{R} \times 7(2)$$
 ...(i)

Now mass defect of N atom = $8 \times 1.008665 + 7 \times 1.007825 - 15.000109 = 0.1239864$ u So binding energy = 0.1239864×931.5 MeV

and mass defect of O atom = $7 \times 1.008665 + 8 \times 1.007825 - 15.003065 = 0.12019044$ u So binding energy = $0.12019044 \times 931.5 \text{ MeV}$

So
$$|B_0 - B_N| = 0.0037960 \times 931.5 \text{ MeV}$$
 ...(ii

from (i) and (ii) we get R = 3.42 fm.

- 2. The ends Q and R of two thin wires, PQ and RS, are soldered (joined) together. Initially each of the wires has a length of 1 m at 10°C. Now the end P is maintained at 10°C, while the end S is heated and maintained at 400 °C. The system is thermally insulated from its surroundings. If the thermal conductivity of wire PQ is twice that of the wire RS and the coefficient of linear thermal expansion of PQ is 1.2×10⁻⁵ K⁻¹, the change in length of the wire PQ is
 - (A) 0.78 mm
- (B) 0.90 mm
- C) 1.56 mm
- (D) 2.34 mm

1

Sol. (A)

From given data;

$$\frac{R_{PQ}}{R_{RS}} = \frac{1}{2}$$

So,
$$T-10=\frac{400-T}{2}$$

$$\Rightarrow$$
 T = 140 °C

As a function of x,

$$T(x) = 10 + 130 x$$

$$\Rightarrow \Delta T(x) = T(x) - 10 = 130x$$

Extension in a small element of length dx is

$$d\ell = \alpha \Delta T(x) dx = 130 \alpha x dx$$

⇒ Net extension

$$\Delta \ell = 130\alpha \int_{0}^{1} x dx = \frac{130}{2} \times 1.2 \times 10^{-5} \times 1$$

or,
$$\Delta \ell = 0.78$$
 mm.

- 3. An accident in a nuclear laboratory resulted in deposition of a certain amount of radioactive material of half-life 18 days inside the laboratory. Tests revealed that the radiation was 64 times more than the permissible level required for safe operation of the laboratory. What is the minimum number of days after which the laboratory can be considered safe for use?
 - (A) 64
- (B) 90
- (C) 108
- (D) 120

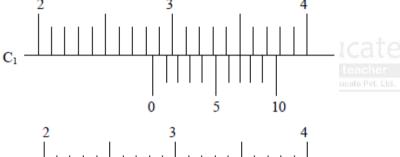
Sol. (C)

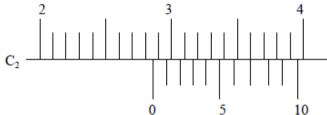
Required activity =
$$\frac{\text{Initial activity}}{64} = \frac{\text{Initial activity}}{2^6}$$

Time required = 6 half lives

$$= 6 \times 18 \text{ days} = 108 \text{ days}$$

4. There are two Vernier calipers both of which have 1 cm divided into 10 equal divisions on the main scale. The Vernier scale of one of the calipers (C₁) has 10 equal divisions that correspond to 9 main scale divisions. The Vernier scale of the other caliper (C₂) has 10 equal divisions that correspond to 11 main scale divisions. The readings of the two calipers are shown in the figure. The measured values (in cm) by calipers C₁ and C₂ respectively, are





- (A) 2.87 and 2.86
- (B) 2.87 and 2.87
- (C) 2.87 and 2.83
- (D) 2.85 and 2.82

Sol. (C)

In first; main scale reading = 2.8 cm.

Vernier scale reading =
$$7 \times \frac{1}{10} = 0.07 \text{ cm}$$

So reading = 2.87 cm;

In second; main scale reading = 2.8 cm

Vernier scale reading =
$$7 \times \frac{-0.1}{10} = \frac{-0.7}{10} = -0.07 \text{ cm}$$

so reading =
$$(2.80 + 0.10 - 0.07)$$
 cm = 2.83 cm

5. A gas is enclosed in a cylinder with a movable frictionless piston. Its initial thermodynamic state at pressure $P_1 = 10^5$ Pa and volume $V_i = 10^{-3}$ m³ changes to a final state at $P_f = (1/32) \times 10^5$ Pa and $V_f = 8 \times 10^{-3}$ m³ in an adiabatic quasi-static process, such that $P^3V^5 =$ consider another thermodynamic process that brings the system from the same initial state to the same final state in

two steps: an isobaric expansion at P_i followed by an isochoric (isovolumetric) process at volume V_f. The amount of heat supplies to the system in the two-step process is approximately

- (A) 112 J
- (B) 294 J
- (C) 588 J
- (D) 813 J

Sol. (C)

$$\gamma = \frac{5}{3} \implies$$
 monoatomic gas

From first law of thermodynamics $H = W + \Delta U$

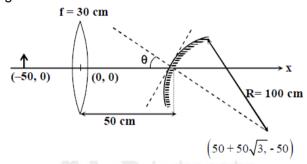
$$W = P_i \Delta V = 700 J$$

$$\Delta U = nC_v \Delta T$$

$$=\frac{3}{2}[P_fV_f-P_iV_i]=-\frac{900}{8}J$$

So,
$$H = W + \Delta U = 588J$$

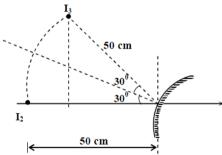
A small object is placed 50 cm to the left of a thin convex lens of focal length 30 cm. A convex 6. spherical mirror of radius of curvature 100 cm is placed to the right of the lens at a distance of 50 cm. The mirror is tilted such that the axis of the mirror is at an angle $\theta = 30^{\circ}$ to the axis of the lens, as shown in the figure.



If the origin of the coordinate system is taken to be at the centre of the lens, the coordinates (in cm) of the point (x, y) at which the image is formed are

- (A) $(25, 25\sqrt{3})$
- (B) $(125/3,25/\sqrt{3})$ (C) $(50-25\sqrt{3},25)$

(A) Sol.



First Image I₁ from the lens will be formed at 75 cm to the right of the lens.

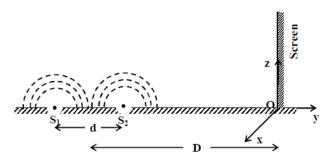
Taking the mirror to be straight, the image I₂ after reflection will be formed at 50 cm to the left of the mirror.

On rotation of mirror by 30° the final image is I₃.

So
$$x = 50 - 50\cos 60^{\circ} = 25 \text{ cm}$$
. and $y = 50 \sin 60^{\circ} = 25\sqrt{3} \text{ cm}$

ONE OR MORE THAN ONE CHOICE CORRECT

While conducting the Young's double slit experiment, a student replaced the two slits with a large 7. opaque plate in the x-y plane containing two small holes that act as two coherent point sources (S₁, S₂) emitting light of wavelength 600 nm. The student mistakenly placed the screen parallel to the x-z plane (for z > 0) at a distance D = 3 m from the mid-point of S_1S_2 , as shown schematically in the figure. The distance between the sources d = 0.6003 mm. The origin O is at the intersection of the screen and the line joining S_1S_2 , which of the following is(are) true of the intensity pattern on the screen?



- (A) Hyperbolic bright and dark bands with foci symmetrically placed about O in the x-direction
- (B) Semi circular bright and dark bands centred at point O
- (C) The region very close to the point O will be dark
- (D) Straight bright and dark bands parallel to the x-axis
- Sol. (B, C

Since S₁S₂ line is perpendicular to screen shape of pattern is concentric semicircle

At
$$O, \frac{2\pi}{\lambda}(S_1O - S_2O) = \frac{2\pi \times 0.6003 \times 10^{-3}}{600 \times 10^{-9}} = 2001\pi$$

- .. darkness close to O.
- 8. In an experiment to determine the acceleration due to gravity g, the formula used for the time period of a periodic motion is $T=2\pi\sqrt{\frac{7(R-r)}{5g}}$. The values of R and r are measured to be (60 ± 1) mm and

 (10 ± 1) mm, respectively. In five successive measurements. The time period is found to be 0.52 s, 0.56s, 0.57 s, 0.54 s and 0.59 s. The least count of the watch used for the measurement of time period is 0.01 s. Which of the following statements(s) is (are true?

- (A) The error in the measurement of r is 10%
- (B) The error in the measurement of T is 3.57%
- (C) The error in the measurement of T is 2%
- (D) The error in the determined value of g is 11%
- Sol. (A, B, D)

Error in T

$$T_{mean} = \frac{0.52 + 0.56 + 0.57 + 0.54 + 0.59}{5} = 0.556 \approx 0.56 \, s$$

$$\Delta T_{mean} = 0.02$$

$$\therefore$$
 error in T is given by $\frac{0.02}{0.56} \times 100 = 3.57\%$

Error in
$$r = \frac{1}{10} \times 100 = 10\%$$

Error in g :
$$T = 2\pi \sqrt{\frac{7(R-r)}{5g}}$$

$$T^2 = 4\pi^2 \frac{7}{5} \left(\frac{R - r}{g} \right)$$

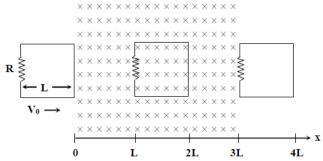
$$g = \frac{28\pi^2}{5} \left(\frac{R - r}{T^2} \right)$$

$$\frac{\Delta g}{g} = \left(\frac{\Delta R + \Delta r}{R - r}\right) + 2\frac{\Delta T}{T} = \frac{2}{50} + 2 \times 0.0357$$

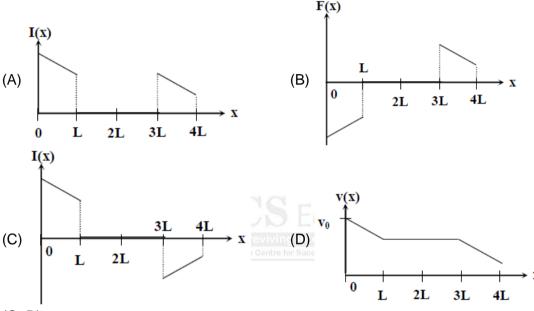
$$\therefore \frac{\Delta g}{g} \times 100 \approx 11\%$$

9. A rigid wire loop of square shape having side of length L and resistance R is moving along the x-axis with a constant velocity v_0 in the plane of the paper. At t = 0, the right edge of the loop enters

a region of length 3L where there is a uniform magnetic field B_0 into the plane of the paper, as shown in the figure. For sufficiently large v_0 , the loop eventually crosses the region. Let x be the location of the right edge of the loop. Let v(x), I(x) and F(x) represent the velocity of the loop, current in the loop, and force on the loop, respectively, as a function of x. Counter-clockwise current is taken as positive.



Which of the following schematic plot(s) is(are) correct? (Ignore gravity)



Sol. (C, D)

For right edge of loop from x = 0 to x = L

$$i = + \frac{vBL}{R}$$

$$F = iLB = \frac{vB^2L^2}{R} \text{ (leftwards)}$$

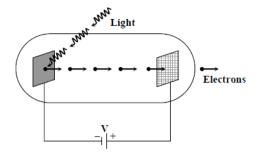
$$-mv\frac{dv}{dx} = \frac{vB^2L^2}{R}$$

$$\therefore V(x) = V_0 - \frac{B^2L^2}{mR}x$$

$$i(x) = \frac{v_0 BL}{R} - \frac{B^3 L^3}{mR^2} x$$

$$F(x) = \frac{v_0 B^2 L^2}{R} - \frac{B^4 L^4}{mR^2} x$$
 (leftwards)

10. Light of wavelength λ_{ph} falls on a cathode plate inside a vacuum tube as shown in the figure. The work function of the cathode surface is ϕ and the anode is a wire mesh of conducting material kept at a distance d from the cathode. A potential difference V is maintained between the electrodes. If the minimum de Broglie wavelength of the electrons passing through the anode is λ_e , which of the following statement(s) is(are) true?



- (A) For large potential difference $(V >> \phi / e), \lambda_e$ is approximately halved if V is made four times
- (B) λ_{e} increases at the same rate as λ_{ph} for λ_{ph} < hc/ φ
- (C) λ_e is approximately halved, if d is doubled
- (D) λ_e decreases with increase in ϕ and λ_{ph}

Sol. (A)

Equation Becomes

$$\frac{hC}{\lambda_{nh}} + eV - \phi = \frac{P_{max}^2}{2m}$$

$$\frac{hC}{\lambda_{ph}} + eV - \varphi = \frac{h^2}{2m\lambda_e^2}$$

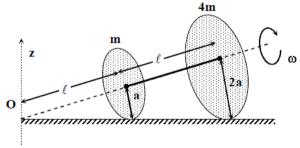
For
$$V \gg \frac{\phi}{e}$$

$$\Rightarrow \ \phi << eV \ \ \text{and} \ \ \frac{hC}{\lambda_{ph}} << eV \ \ \Rightarrow \ \ eV = \frac{h^2}{2m\lambda_e^2}$$

$$\lambda_e \alpha \frac{1}{\sqrt{V}}$$

when V is made four times λ_e is halved.

11. Two thin circular discs of mass m and 4m, having radii of a and 2a, respectively, are rigidly fixed by a massless, rigid rod of length $\ell = \sqrt{24}a$ through their centers. This assembly is laid on a firm and flat surface and set rolling without slipping on the surface so that the angular speed about the axis of the rod is ω . The angular momentum of the entire assembly about the point 'O' is \vec{L} (see the figure). Which of the following statement(s) is (are) true?



- (A) The magnitude of angular momentum of the assembly about its center of mass is 17 ma 2 $_{\odot}/$ 2
- (B) The magnitude of the z-component of \vec{L} is 55 ma $^2\omega$
- (C) The magnitude of angular momentum of center of mass of the assembly about the point O is 81 $\text{ma}^2\omega$
- (D) The center of mass of the assembly rotates about the z-axis with an angular speed of $\,\omega\,/\,5\,$

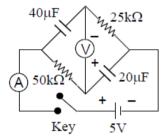
Sol. (D) or (A, D)

$$\omega_{z} = \frac{\omega a}{\ell} \cos \theta = \omega / 5$$

- 12. Consider two identical galvanometers and two identical resistors with resistance R. If the internal resistance of the galvanometers $R_G < R/2$, which of the following statement(s) about any one of the galvanometers is(are) true?
 - (A) The maximum voltage range is obtained when all the components are connected in series
 - (B) The maximum voltage range is obtained when the two resistors and one galvanometer are connected in series, and the second galvanometer is connected in parallel to the first galvanometer
 - (C) The maximum current range is obtained when all the components are connected in parallel
 - (D) The maximum current range is obtained when the two galvanometers are connected in series and the combination is connected in parallel with both the resistors
- Sol. (A, C)

For maximum voltage range across a galvanometer, all the elements must be connected in series. For maximum current range through a galvanometer, all the elements should be connected in parallel.

13. In the circuit shown below, the key is pressed at time t = 0. Which of the following statement(s) is (are) true?



- (A) The voltmeter displays 5 V as soon as the key is pressed, and displays +5 V after a long time
- (B) The voltmeter will display 0 V at time $t = \ell n \ 2$ seconds
- (C) The current in the ammeter becomes 1/e of the initial value after 1 second
- (D) The current in the ammeter becomes zero after a long time
- Sol. (A, B, C, D)

at t = 0, voltage across each capacitor is zero, so reading of voltmeter is -5 Volt.

at $\,t=\infty$, capacitors are fully charged. So for ideal voltmeter, reading is 5 Volt.

at transient state.

$$I_1 = \frac{5}{50} e^{-\frac{t}{\tau}} \text{mA}, I_2 = \frac{5}{25} e^{-\frac{t}{\tau}} \text{ and } I = I_1 + I_2$$

where $\tau = 1$ sec

So I becomes 1/e times of the initial current after 1 sec.

The reading of voltmeter at any instant $\Delta V_{40\mu F} - \Delta V_{50k\Omega} = 5 \left(1 - e^{-\frac{t}{\tau}}\right) - 5e^{-\frac{t}{\tau}}$

So at $t = \ell n2$ sec, reading of voltmeter is zero.

- 14. A block with mass M is connected by a massless spring with stiffness constant k to a rigid wall and moves eithout friction on a horizontal surface. The block oscillates with small amplitude A about an equilibrium position x_0 . Consider two cases: (i) when the block is at $x = x_0 + A$. In both the cases, a particle with mass m (<M) is softly placed on the block after which they stick to each other. Which of the following statement(s) is (are) true about the motion after the mass m is placed on the mass M?
 - (A) The amplitude of oscillation in the first case changes by a factor of $\sqrt{\frac{M}{m+M}}$, whereas in the second case it remains unchanged
 - (B) The final time period of oscillation in both the cases is same
 - (C) The total energy decreases in both the cases
 - (D) The instantaneous speed at x_0 of the combined masses decreases in both the cases

Case (i):
$$\omega' = \sqrt{\frac{k}{M+m}}$$

$$MA\sqrt{\frac{k}{M}} = (M+m)A'\sqrt{\frac{k}{M+m}}$$
, so $A' = A\sqrt{\frac{M}{M+m}}$

$$E' = \frac{1}{2}(M+m)\frac{k}{M+m}A^2\frac{M}{M+m} = \frac{1}{2}\frac{kA^2M}{M+m}$$

$$v' = \frac{Mv}{M+m}$$

Case (ii):
$$\omega' = \sqrt{\frac{k}{M+m}}$$

A remains same

$$E' = \frac{1}{2}(M+m)\frac{k}{M+m}A^2 \text{ (Remains Same)}$$

$$v' = A\sqrt{\frac{k}{M+m}}$$

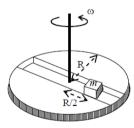
Paragraph-1

A frame of reference that is accelerated with respect to an inertial frame of reference is called a noninertial frame of reference. A coordinate system fixed on a circular disc rotating about a fixed acis with a constant angular velocity ω is an example of a non-inertial frame of reference. The relationship between the force \vec{F}_{rot} experienced by a particle of mass m moving on the rotating disc and the force \vec{F}_{in} experienced by the particle in an inertial frame of reference is

$$\vec{F}_{rot} = \vec{F}_{in} + 2m(\vec{v}_{rot} \times \vec{\omega}) + m(\vec{\omega} \times \vec{r}) \times \vec{\omega}$$

where \vec{v}_{rot} is the velocity of the particle in the rotating frame of reference and \vec{r} is the position vector of the particle with respect to the centre of the disc.

Now consider a smooth slot along a diameter of a disc of radius R rotating counter-clockwise with a constant angular speed ω about its vertical axis through its center. We assign a coordinate system with the origin at the center of the disc, the x-axis along the slot, the y-axis perpendicular to the slot and the z-axis along the rotation axis $(\vec{\omega} = \omega \hat{k})$. A small block of mass m is gently placed in the slot at $\vec{r} = (R/2)\hat{i}$ at t = 0 and is constrained to move only along the slot.



15. The distance r of the black at time t is

(A)
$$\frac{R}{4} \left(e^{2\omega t} + e^{-2\omega t} \right)$$

(B)
$$\frac{R}{2}\cos 2\omega t$$

(C)
$$\frac{R}{2}\cos\omega$$

(A)
$$\frac{R}{4} \left(e^{2\omega t} + e^{-2\omega t} \right)$$
 (B) $\frac{R}{2} \cos 2\omega t$ (C) $\frac{R}{2} \cos \omega t$ (D) $\frac{R}{4} \left(e^{\omega t} + e^{-\omega t} \right)$

Sol.

 $v \frac{dv}{dr} = \omega^2 r$, where v is the velocity of the block radially outward

$$\int_0^v v \, dv = \omega^2 \int_{R/2}^r r \, dr$$

$$\Rightarrow \ v = \omega \sqrt{r^2 - \frac{R^2}{4}}$$

$$\begin{split} &\int_{R/2}^{r} \frac{dr}{\sqrt{r^2 - \frac{R^2}{4}}} = \omega \int_{0}^{t} dt \\ &r = \frac{R}{4} \Big(e^{\omega t} + e^{-\omega t} \Big) \end{split}$$

16. The net reaction of the disc on the block is

(C)
$$\frac{1}{2}$$
m ω^2 R $\left(e^{wt} - e^{-\omega t}\right)\hat{j} + mg\hat{k}$

(A) $-m\omega^2 R \cos \omega t \hat{j} - mg\hat{k}$

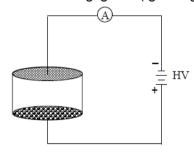
(D)
$$\frac{1}{2}$$
m ω^2 R $\left(e^{2wt}-e^{-2\omega t}\right)\hat{j}+mg\hat{k}$

(B) $-m\omega^2 R \sin \omega t \hat{i} - m\alpha \hat{k}$

$$\begin{split} \vec{F}_{\text{rot}} &= \vec{F}_{\text{in}} + 2m \left(\vec{v}_{\text{rot}} \times \vec{\omega} \right) + m \left(\vec{\omega} \times \vec{r} \right) \times \vec{\omega} \\ &= -m\omega^2 r \hat{i} + 2m v_{\text{rot}} \omega \left(-\hat{j} \right) + m\omega^2 r \hat{i} \\ &= -2m v_{\text{rot}} \omega \hat{j} \\ v_{\text{rot}} &= \frac{dr}{dt} = \frac{\omega R}{4} \left(e^{\omega t} - e^{-\omega t} \right) \\ \vec{F}_{\text{rot}} &= -\frac{m\omega^2 R}{2} \left(e^{\omega t} - e^{-\omega t} \right) \hat{j} \\ \vec{F}_{\text{net}} &= -\vec{F}_{\text{rot}} + mg \, \hat{k} \\ &= \frac{m\omega^2 R}{2} \left(e^{\omega t} - e^{-\omega t} \right) \hat{j} + mg \hat{k} \end{split}$$

PARAGRAPH 2

Consider an evacuated cylindrical chamber of height h having rigid conducting plates at the ends and an insulating curved surface as shown in the figure. A number of spherical balls made of a light weight and soft material and coated with a conducting material are placed on the bottom plate. The balls have a radius r << h. Now a high voltage source (HV) is connected across the conducting plates such that the bottom plate is at $+V_0$ and the top plate at $-V_0$. Due to their conducting surface, the balls will get charged, will become equipotential with the plate and are repelled by it. The balls will eventually collide with the top plate, where the coefficient of restitution can be taken to be zero due to the soft nature of the material of the balls. The electric field in the chamber can be considered to be that of a parallel plate capacitor. Assume that there are no collisions between the balls and the interaction between them is negligible. (Ignore gravity)



17. Which one of the following statements is correct?

- (A) The balls will bounce back to the bottom plate carrying the opposite charge they went up with
- (B) The balls will execute simple harmonic motion between the two plates
- (C) The balls will bounce back to the bottom plate carrying the same charge they went up with
- (D) The balls will stick to the top plate and remain there

Sol. (A)

After hitting the top plate, the balls will get negatively charged and will now get attracted to the bottom plate which is positively charged. The motion of the balls will be periodic but not SHM.

18. The average current in the steady state registered by the ammeter in the circuit will be

(A) proportional to $V_0^{1/2}$

(C) proportional to the potential $\,V_0\,$

(D) zero

Sol.

If Q is charge on balls, then Q $\alpha\,V_0$

Also $h = \frac{1}{2}at^2 = \frac{1}{2}\left(\frac{QV_0}{mh}\right)t^2$

$$\Rightarrow \ t\alpha \frac{1}{V_0}$$

Now,
$$I_{av} \alpha \frac{Q}{t}$$

$$\Rightarrow I_{av} \alpha V_0^2$$

- (B) proportional to V_0^2

...(i)



PART II: CHEMISTRY

19. The correct order of acidity for the following compounds is

$$(A) I > II > III > IV$$

(B)
$$III > I > II > IV$$

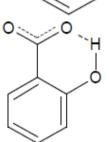
(C)
$$III > IV > II > I$$

(D)
$$I > III > IV > II$$

Sol. (A)

Stabler the conjugate base stronger the acid

Conjugate base stabilized by intramolecular H-bond from both the sides



Conjugate base stabilized by intramolecular H-bond from one side.

- 20. The geometries of the ammonia complexes of Ni²⁺,Pt²⁺ and Zn²⁺, respectively, are
 - (A) octahedral, square planar and tetrahedral
 - (B) square planar, octahedral and tetrahedral
 - (C) tetrahedral, square planar and octahedral
 - (D) octahedral, tetrahedral and square planar

Sol. (A)

$$\begin{bmatrix} \text{Ni}\big(\text{NH}_3\big)_6 \end{bmatrix}^{2+}; \ \begin{bmatrix} \text{Pt}\big(\text{NH}_3\big)_4 \end{bmatrix}^{2+}; \begin{bmatrix} \text{Zn}\big(\text{NH}_3\big)_4 \end{bmatrix}^{2+} \\ \text{octahedral} \end{bmatrix}^{2+}$$

21. For the following electrochemical cell at 298 K,

$$E_{cell} = 0.092V$$

when
$$\frac{\left[M^{2+}(aq)\right]}{\left[M^{4+}(aq)\right]} = 10^{x}$$

Given:
$$E_{M^{4+}/M^{2+}}^0 = 0.151V$$
; $2.303 \frac{RT}{F} = 0.059V$

The value of x is

$$(A) -2$$

$$(B) -1$$

$$(C)$$
 1

11

Sol. (D)

$$H_2 - 2e \square 2H^+$$

$$M^{4+} + 2e \square M^{2+}$$

Net cell reaction : $H_2 + M^{4+} \square 2H^+ + M^{2+}$

$$\boldsymbol{E}_{\text{cell}} = \boldsymbol{E}_{\text{cell}}^{0} - \frac{0.059}{2} log \frac{\left[\boldsymbol{H}^{+}\right] \left[\boldsymbol{M}^{2+}\right]}{\left[\boldsymbol{M}^{4+}\right] \times \boldsymbol{P}_{\boldsymbol{H}_{2}}}$$

$$0.092 = 0.151 - \frac{0.059}{2} \log 10^{x}$$

$$0.092 = 0.151 - \frac{0.059}{2} \times \frac{0.059 \times 2}{2} = 0.151 - 0.092$$
$$0.059 \times = 0.059 \times 2$$

X = 2

22. The major product of the following reaction sequence is

23. In the following reaction sequence in aqueous solution, the species X, Y and Z, respectively, are

$$S_2O_3^{2-} \xrightarrow{Ag^+} \underset{\text{solution}}{X} \xrightarrow{Ag^+} \underset{\text{precipitate}}{Y} \xrightarrow{\text{withtime}} \underset{\text{precipitate}}{Z}$$

(A) $[Ag(S_2O_3)_2]^{3-}$, $Ag_2S_2O_3$, Ag_2 , S

(B)
$$[Ag(S_2O_3)_3]^{5-}$$
, Ag_2SO_3 , Ag_2S

(C) $[Ag(SO_3)_2]^{3-}$, $Ag_2S_2O_3$, Ag

(D)
$$[Ag(SO_3)_3]^{3-}$$
, Ag_2SO_4 , Ag_3O_4

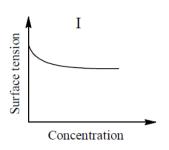
Sol. (A)

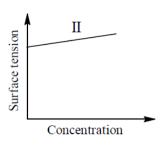
Sol.

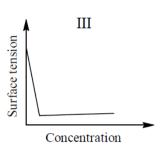
$$\begin{array}{c} \mathsf{Ag^{\scriptscriptstyle +}} + 2\mathsf{S}_2\mathsf{O}_3^{2\scriptscriptstyle -} \to & \left[\mathsf{Ag}(\mathsf{S}_2\mathsf{O}_3)_2\right]^{3\scriptscriptstyle -} \xrightarrow{} & \mathsf{Ag^{\scriptscriptstyle +}} \\ \mathsf{X}_{\phantom{\mathsf{C}}\phantom{\mathsf{C}\phantom{\mathsf{C}\phantom{\mathsf{C}}\phantom{\mathsf{C}}\phantom{\mathsf{C}}\phantom{\mathsf{C}}\phantom{\mathsf{C}}\phantom{\mathsf{C}}\phantom{\mathsf{C}}\phantom{\mathsf{C}}\phantom{\mathsf{C}}\phantom{\mathsf{C}}\phantom{\mathsf{C}}\phantom{\mathsf{C}}\phantom{\mathsf{C}}\phantom{\mathsf{C}\phantom{\mathsf{C}\phantom{\mathsf{C}}\phantom{\mathsf{C}}\phantom{\mathsf{C}}\phantom{\mathsf{C}}\phantom{\mathsf{C}}\phantom{\mathsf{C}}\phantom{\mathsf{C}}\phantom{\mathsf{C}}\phantom{\mathsf{C}}\phantom{\mathsf{C}}\phantom{\mathsf{C}}\phantom{\mathsf{C}}\phantom{\mathsf{C}\phantom{\mathsf{C}}\phantom{\mathsf{C}\phantom{\mathsf{C}}\phantom{\mathsf{C}\phantom{\mathsf{C}}\phantom{\mathsf{C}\phantom{\mathsf{C}}\phantom{\mathsf{C}\phantom{\mathsf{C}}\phantom{\mathsf{C}\phantom{\mathsf{C}}\phantom{\mathsf{C}}\phantom{\mathsf{C}\phantom{\mathsf{C}}\phantom{\mathsf{C}}\phantom{\mathsf$$

 $Ag_2S_2O_3 + H_2O \rightarrow Ag_2S + H_2SO_4$

24. The qualitative sketches I, II and III given below show the variation of surface tension with molar concentration of three different aqueous solutions of KCI, CH₃OH and CH₃OSO₃⁻Na⁺ at room temperature. The correct assignment of the sketches is







(A) I: KCI

II: CH₃OH

III: CH₃(CH₂)₁₁OSO₃-Na⁺

(B) I: CH₃(CH₂)₁₁OSO₃-Na⁺

II: CH₃OH

III: KCI

(C) I : KCI

II: CH₃(CH₂)₁₁OSO₃Na⁺

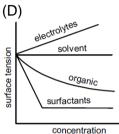
III: CH₃OH

(D) I: CH₃OH

II: KCI

III: CH₃(CH₂)₁₁OSO₃Na⁺





Strong electrolytes like KCI increase the surface tension slightly. Low molar mass organic compounds usually decrease the surface tension. Surface active organic compounds like detergents sharply decrease surface tension

ONE OR MORE THAN ONE CHOICE CORRECT

25. For 'invert sugar', the correct statemet(s) is (are)

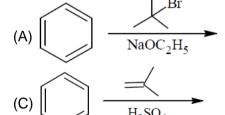
(Given: specific rotations of (+)-sucrose, (+)-maltose, L-(-)-glucose and L-(+)-fructose in aqueous solution are $+66^{\circ}$, $+140^{\circ}$, -52° and $+92^{\circ}$, respectively)

- (A) 'invert sugar' is prepared by acid catalyzed hydrolysis of maltose
- (B) 'invert sugar' is an equimolar micture of D-(+)-glucose and D-(-)-fructose
- (C) specific rotation of 'invert sugar' is -20°
- (D) on reaction with Br₂ water, 'invert sugar' forms saccharic acid as one of the products
- Sol. (B, C)

$$C_{12}H_{22}O_{11}+H_2O \xrightarrow{\quad H^+ \quad} C_6H_{12}O_6 + C_6H_{12}O_6 \\ \xrightarrow[+25^o]{\quad D-(-)-frctose} \\ \xrightarrow[-92^o]{\quad H^+ \quad} C_6H_{12}O_6 + C_6H_{12}O_6$$

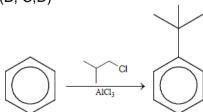
 $\alpha_{\text{invert sugar}} = \frac{+52^{\circ} - 92^{\circ}}{2} = -20^{\circ}$ (average is taken as both monomers are one mole each)

26. Among the following, reaction(s) which gives (give) tert-butyl benzene as the major product is(are)

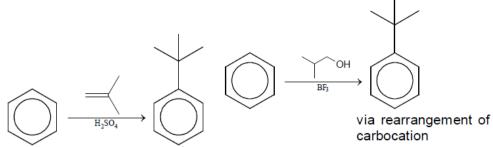


(D)
$$BF_3.OEt_2$$

Sol. (B, C,D)



via rearrangement of carbocation



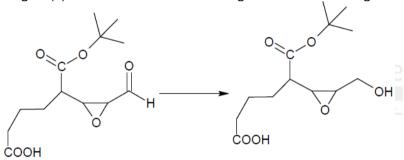
- 27. Extraction of copper from copper pyrite (CuFeS₂) involves
 - (A) crushing followed by concentration of the ore by froth flotation
 - (B) removal of iron as slag
 - (C) self-reduction step to produce 'blister copper' following evolution of SO₂
 - (D) refining of 'blister copper' by carbon reduction
- Sol. (A, B, C)

Refining of blister copper is done by poling technique

- 28. The CORRECT statement(s) for cubic close packed (ccp) three dimensional structure is(are)
 - (A) The number of the nearest neighbours of an atom present in the topmost layer is 12
 - (B) The efficiency of atom packing is 74%
 - (C) The number of octahedral and tetrahedral voids per atom are 1 and 2, respectively
 - (D) The unit cell edge length is $2\sqrt{2}$ times the radius of the atom
- Sol. (B, C, D)

The middle layers will have 12 nearest neighbours. The top-most layer will have 9 nearest neighbours. $4r = a\sqrt{2}$, where 'a' is edge length of unit cell and 'r' is radius of atom.

29. Regent(s) which can be used to bring about the following transformation is (are)



(A) LiAlH₄ in $(C_2H_5)_2O$

(B) BH₃ in THF

(C) NaBH₄ in C₂H₅OH

(D) Raney Ni/H2 in THF

Sol. (C, D)

NaBH₄ and Raney Ni/H₂ do not react with acid, ester or epoxide entities of an organic compound

- 30. Mixture(s) showing positive deviation from Raoult's law at 35°C is (are)
 - (A) carbon tetrachloride + methanol
- (B) carbon disulphide + acetone

(C) benzene + toluene

(D) phenol + aniline

Sol. (A, B)

Benzene + toluene will form ideal solution.

Phenol + aniline will show negative deviation.

- 31. The nitrogen containing compound produced in the reaction of HNO₃ with P₄O₁₀
 - (A) can also be prepared by reaction of P₄ and HNO₃
 - (B) is diamagnetic
 - (C) contains one N N bond
 - (D) reacts with Na metal producing a brown gas
- Sol. (B, D

$$2NHO_{3} \xrightarrow{P_{4}O_{10}} N_{2}O_{5}$$

$$O \qquad O \qquad O$$

$$O \qquad$$

- 32. According to Molecular Orbital Theory
 - (A) C₂²⁻ is expected to be diamagnetic
 - (B) O_2^{2+} expected to have a longer bond length than O_2
 - (C) N₂ and N₂ have the same bond order
 - (D) $\mathrm{He}_{2}^{\scriptscriptstyle +}$ has the same energy as two isolated He atoms

Sol. (A. C)

Integer type

PARAGRAPH 1

Thermal decomposition of gaseous X₂ to gaseous X at 298 K takes place according to the following equation: $X_2(g) \square 2X(g)$

The standard reaction Gibbs energy, $\Delta_r G^0$, of this reaction is positive. At the start of the reaction, there is one mole of X₂ and no X. As the reaction proceeds, the number of moles of X formed is given by β . Thus, $\beta_{equilibrium}$ is the number of moles of X formed at equilibrium. The reaction is carried out at a constant total pressure of 2 bar. Consider the gases to behave ideally. (Given: R = 0.083 L bar K⁻¹mol⁻¹)

33. The equilibrium constant K_p for this reaction at 298 K, in term of $\beta_{equilibrium}$ is

(A)
$$\frac{8\beta_{\text{equilibrium}}^2}{2 - \beta_{\text{equilibrium}}}$$

(B)
$$\frac{8\beta_{\text{equilibrium}}^2}{4 - \beta_{\text{equilibrium}}^2}$$

$$\text{(A)} \ \frac{8\beta_{\text{equilibrium}}^2}{2-\beta_{\text{equilibrium}}} \qquad \qquad \text{(B)} \ \frac{8\beta_{\text{equilibrium}}^2}{4-\beta_{\text{equilibrium}}^2} \qquad \qquad \text{(C)} \ \frac{4\beta_{\text{equilibrium}}^2}{2-\beta_{\text{equilibrium}}} \qquad \qquad \text{(D)} \ \frac{4\beta_{\text{equilibrium}}^2}{4-\beta_{\text{equilibrium}}^2} \qquad \qquad \text{$$

(D)
$$\frac{4\beta_{\text{equilibrium}}^2}{4 - \beta_{\text{equilibrium}}^2}$$

Sol.

$$X_2(g) \square 2x(g)$$

$$1-\frac{\beta_e}{2}$$

Total number of moles at equilibrium

$$\Rightarrow 1 - \frac{\beta_e}{2} + \beta_e$$

$$\Rightarrow 1 + \frac{\beta_{\epsilon}}{2}$$

$$K_{p} = \frac{(p_{x})^{2}}{p_{x_{2}}} = \frac{\left(\frac{\beta_{e} \times 2}{1 + \frac{\beta_{e}}{2}}\right)}{1 + \frac{\beta_{e}}{2}} = \frac{2\beta_{e}^{2}}{1 - \frac{\beta_{e}^{2}}{4}}$$

$$K_p = \frac{8\beta_e^2}{4 - \beta_e^2}$$

- The INCORRECT statement among the following, for this reaction is 34.
 - (A) Decrease in the total pressure will result in formation of more moles of gaseous X
 - (B) At the start of the reaction, dissociation of gaseous X2 takes place spontaneously
 - (C) $\beta_{equilibrium} = 0.7$
 - (D) $K_c < 1$
- Sol.

There is no data given to find the $\beta_{\text{equilibrium}}$ exact value.

$$\Delta G_c^0 = -2.303 \ RT \ log \ K_c$$

$$log K_c = -1$$

$$K_c < 1$$

PARAGRAPH 2

Treatment of compound O with KMnO₄/H⁺ gave P, which on heating with ammonia gave Q. The compound Q on treatment with Br₂/NaOH produced R. On strong heating, Q gave S, which on further treatment with ethyl 2-bromopropanoate in the presence of KOH followed by acidification, gave a compound T.

35. The cpmpound R is

(A) Sol.

The compound T is 36.

(A) glycine

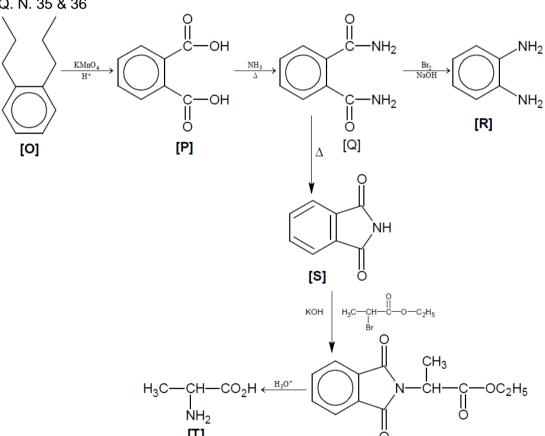
(B) alanine

(C) valine

(D) serine

Sol. (B)

Q. N. 35 & 36



(alanine)

PART III: MATHEMATICS

$$P^{50} - Q = I, \text{ then } \frac{q_{31} + q_{32}}{q_{21}} \text{ equals}$$

Sol.

$$P\begin{bmatrix} 1 & 0 & 0 \\ 4 & 1 & 0 \\ 16 & 4 & 1 \end{bmatrix} = \begin{bmatrix} 0 & 0 & 0 \\ 4 & 0 & 0 \\ 16 & 4 & 0 \end{bmatrix} + \begin{bmatrix} 1 & 0 & 0 \\ 0 & 1 & 0 \\ 0 & 0 & 1 \end{bmatrix}$$

Let
$$A = \begin{bmatrix} 0 & 0 & 0 \\ 4 & 0 & 0 \\ 16 & 4 & 0 \end{bmatrix} \Rightarrow A^2 = \begin{bmatrix} 0 & 0 & 0 \\ 0 & 0 & 0 \\ 16 & 0 & 0 \end{bmatrix}$$
 and $A^3 = \begin{bmatrix} 0 & 0 & 0 \\ 0 & 0 & 0 \\ 0 & 0 & 0 \end{bmatrix}$

 \Rightarrow Aⁿ is a null matrix \forall n \geq 3

$$P^{50} = (I + A)^{50} = I + 50A + \frac{50 \times 49}{2}A^2$$

$$Q + I = \begin{bmatrix} 1 & 0 & 0 \\ 0 & 1 & 0 \\ 0 & 0 & 1 \end{bmatrix} + 50 \begin{bmatrix} 0 & 0 & 0 \\ 4 & 0 & 0 \\ 16 & 4 & 0 \end{bmatrix} + 25 \times 49 \begin{bmatrix} 0 & 0 & 0 \\ 0 & 0 & 0 \\ 16 & 0 & 0 \end{bmatrix}$$

$$\Rightarrow \left(\frac{q_{31} + q_{32}}{q_{21}}\right) = \frac{16(50 + 25 \times 49) + 50 \times 4}{50 \times 4}$$

$$16 \times 51 + 8$$

$$=\frac{16\times51+8}{8}=102+1=103$$

$$=\frac{16\times51+8}{8}=102+1=103$$

38. Area of the region
$$\{(x,y) \in \square^2 : y \ge \sqrt{|x+3|}, 5y \le x+9 \le 15\}$$
 is equal to

(A)
$$\frac{1}{6}$$

(B)
$$\frac{4}{3}$$

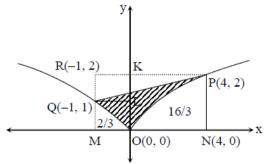
(B)
$$\frac{4}{3}$$
 (C) $\frac{3}{2}$

(D)
$$\frac{5}{3}$$

Sol.

Shifting origin to (-3, 0)

Area
$$\{(x,y) \in \mathbb{R}^2 : y \ge \sqrt{|x|}, 5y \le x + 6 \le 15 \}$$



Area = Region (OPK) + Region (QLKR) + Region (OLQ) - Triangle (PQR)

Area
$$\frac{8}{3} + 1 + \frac{1}{3} - \frac{5}{2} = \frac{3}{2}$$

39. The value of
$$\sum_{k=1}^{13} \frac{1}{\sin\left(\frac{\pi}{4} + \frac{(k-1)\pi}{6}\right) \sin\left(\frac{\pi}{4} + \frac{k\pi}{6}\right)}$$
 is equal to

(A)
$$3 - \sqrt{3}$$

(B)
$$2(3-\sqrt{3})$$

(C)
$$2(\sqrt{3}-1)$$

(B)
$$2(3-\sqrt{3})$$
 (C) $2(\sqrt{3}-1)$ (D) $2(2+\sqrt{3})$

Sol.

$$T_{k} = 2 \left(\cot \left(\frac{\pi}{4} + (k-1)\frac{\pi}{6} \right) - \cot \left(\frac{\pi}{4} + \frac{k\pi}{6} \right) \right)$$

$$\boldsymbol{T_k} = \boldsymbol{V_{k-1}} - \boldsymbol{V_k}$$

$$S_{13} = V_0 - V_{13}$$

$$=2\left(\cot\left(\frac{\pi}{4}\right)-\cot\left(\frac{\pi}{4}+\frac{13\pi}{6}\right)\right)$$

$$=2\left(1-\cot\frac{5\pi}{12}\right)$$

$$=2(1-(2-\sqrt{3}))=2(\sqrt{3}-1)$$

40. Let $b_i > 1$ for i = 1, 2,, 101. Suppose $log_e b_1, log_e b_2, ..., log_e b_{101}$ are in Arithmetic Progression (A.

P.) with the common difference $log_e 2$. Suppose $a_1, a_2, ..., a_{101}$ are in A. P. such that $a_1 = b_1$ and $a_{51} = b_{51}$. If $t = b_1 + b_2 + + b_{51}$ and $s = a_1 + a_2 + ... + a_{51}$, then

(A)
$$s > t$$
 and $a_{101} > b_{101}$

(B) s > t and
$$a_{101} < b_{101}$$

(C)
$$s < t$$
 and $a_{101} > b_{101}$

(D)
$$s < t$$
 and $a_{101} < b_{101}$

Sol. (B)

> a₂, a₃ a₅₀ are Arithmetic Means and b₂, b₃,, b₅₀ are Geometric Means between a₁(= b₁) and $a_{51} (= b_{51})$

Hence $b_2 < a_2, b_3 < a_3$

$$\Rightarrow$$
 t < S

Also a₁, a₅₁, a₁₀₁ is an Arithmetic Progression and b₁, b₅₁, b₁₀₁ is a Geometric Progression

Since $a_1 = b_1$ and $a_{51} = b_{51}$

$$\Rightarrow b_{101} > a_{101}$$

The value of $\int_{\pi}^{\frac{\pi}{2}} \frac{x^2 \cos x}{1 + e^x} dx$ is equal to 41.

(A)
$$\frac{\pi^2}{4} - 2$$
 (B) $\frac{\pi^2}{4} + 2$

(B)
$$\frac{\pi^2}{4} + 2$$

(C)
$$\pi^2 - e^{\frac{\pi^2}{2}}$$

(C)
$$\pi^2 - e^{\frac{\pi}{2}}$$
 (D) $\pi^2 + e^{\frac{\pi}{2}}$

Sol.

$$= \int\limits_0^{\pi/2} \Biggl(\frac{x^2 \cos x}{1 + e^x} + \frac{x^2 \cos x}{1 + e^{-x}} \Biggr) dx$$

$$= \int\limits_0^{\pi/2} \frac{x^2 \cos x + x^2 e^x \cos x}{1 + e^x} dx = \int\limits_0^{\pi/2} x^2 \cos x \, dx$$

$$= (x^2 \sin x)_0^{\pi/2} - \int_0^{\pi/2} 2x \sin x \, dx$$

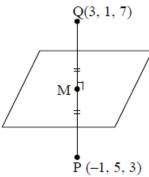
$$= \frac{\pi^2}{4} - 2 \left[\left[\left(x \left(-\cos x \right) \right) \right]_0^{\pi/2} - \int_0^{\pi/2} -\cos x \, dx \right]$$

$$=\frac{\pi^2}{4}-2[-(0-0)+(\sin x)]_0^{\pi/2}$$

$$=\frac{\pi^2}{4}-2$$

- Let P be the image of the point (3, 1, 7) with respect to the plane x y + z = 3. Then the equation of 42. the plane passing through P and containing the straight line $\frac{x}{1} = \frac{y}{2} = \frac{z}{1}$ is
 - (A) x + y 3z = 0
- (B) 3x + z = 0
- (C) x-4y+7z=0 (D) 2x-y=0

Sol. (C)



Mirror image of (3, 1, 7)

$$\frac{x-3}{1} = \frac{y-1}{-1} = \frac{z-7}{1} = \frac{-2(3-1+7-3)}{3}$$

Equation of plane passing through line and (-1, 5, 3)

$$\vec{n} = \begin{vmatrix} x & y & z \\ -1 & 5 & 3 \\ 1 & 2 & 1 \end{vmatrix}$$

$$x - 4y + 7z = 0$$

ONE OR MORE THAN ONE CHOICE CORRECT

- Let a, $b \in \Box$ and $f:\Box \to \Box$ be defined by $f(x) = a\cos(|x^3 x|) + b|x|\sin(|x^3 + x|)$. Then f is 43.
 - (A) differentiable at x = 0 if a = 0 and b = 1
 - (B) differentiable at x = 1 if a = 1 and b = 0 ng internal teacher
 - (C) NOT differentiable at x = 0 if a = 1 and b = 0
 - (D) NOT differentiable at x = 1 if a = 1 and b = 1
- Sol.

$$f(x) = a\cos(x^3 - x) + bx\sin(x(x^2 + 1))$$

It is a differentiable function $\forall x \in R$

- $\text{Let } f(x) = \lim_{n \to \infty} \left(\frac{n^n (x+n) \bigg(x + \frac{n}{2} \bigg) ... \bigg(x + \frac{n}{n} \bigg)}{n! \Big(x^2 + n^2 \Big) \bigg(x^2 + \frac{n^2}{4} \bigg) ... \bigg(x^2 + \frac{n^2}{n^2} \bigg)} \right)^n \text{, for all } x > 0. \text{ Then}$ 44.

- (A) $f\left(\frac{1}{2}\right) \ge f(1)$ (B) $f\left(\frac{1}{3}\right) \le f\left(\frac{2}{3}\right)$ (C) $f'(2) \le 0$ (D) $\frac{f'(3)}{f(3)} \ge \frac{f'(2)}{f(2)}$

Sol.

$$f(x) = \lim_{n \to \infty} \left(\frac{\prod_{r=1}^{n} \left(1 + \frac{rx}{n} \right)}{\prod_{r=1}^{n} \left(1 + \left(\frac{rx}{n} \right)^{2} \right)} \right)$$

$$= e^{\int\limits_{0}^{1} \ln(1+xy) dy - \ln(1+(xy)^2 dy} = e^{\int\limits_{0}^{x} \ln(1+t) dt - \ln\left(1+t^2\right) dt}$$

$$f'(x) = f(x) ln \left(\frac{1+x}{1+x^2} \right)$$

for $x \in (0,1)$ it is increasing function

$$f'(2) = f(2) \ln \left(\frac{3}{5}\right) < 0$$

$$\frac{f'(3)}{f(3)} = ln\left(\frac{2}{5}\right), \frac{f'(2)}{f(2)} = ln\left(\frac{3}{5}\right)$$

- 45. Let $f: \Box \to (0, \infty)$ and $g: \Box \to \Box$ be twice differentiable functions such that f' and g'' are continuous functions on \Box . Suppose $g: \Box \to f'(2) = g(2) = 0$, $f''(2) \neq 0$ and $g'(2) \neq 0$. If $\lim_{x \to 2} \frac{f(x)g(x)}{f'(x)g'(x)} = 1$, then
 - (A) f has a local minimum at x = 2
- (B) f has a local maximum at x = 2

(C) f''(2) > f(2)

(D) f(x) - f''(x) = 0 for at least one $x \in \square$

Sol. (A, D)

$$\lim_{x\to 2} \frac{f(x)g(x)}{f'(x)g'(x)}$$

$$\Rightarrow \lim_{x \to 2} \frac{f'(x)g(x) + g'(x)f(x)}{f''(x)g'(x) + f'(x)g''(x)}$$

$$\Rightarrow \frac{g'(2)f(2)}{f''(2)g'(2)} = 1$$

$$\Rightarrow$$
 f(2) = f "(2)

Since
$$f(2) > 0, f''(2) > 0$$

- \Rightarrow f has a local minimum at x = 2
- 46. Let $\hat{\mathbf{u}} = \mathbf{u}_1 \hat{\mathbf{i}} + \mathbf{u}_2 \hat{\mathbf{j}} + \mathbf{u}_3 \hat{\mathbf{k}}$ be a unit vector is \Box 3 and $\mathbf{w} = \frac{1}{\sqrt{6}} (\hat{\mathbf{i}} + \hat{\mathbf{j}} + 2\hat{\mathbf{k}})$. Given that there exists a vector

 \vec{v} in \Box 3 such that $|\hat{u} \times \vec{v}| = 1$ and $\hat{w}.(\hat{u} \times \vec{v}) = 1$. Which of the following statement(s) is (are correct?

- (A) There is exactly one choice for such \vec{v} (B) There are infinitely many choices for such \vec{v}
- (C) If $\hat{\mathbf{u}}$ lies in the xy-plane then $|\mathbf{u}_1| = |\mathbf{u}_2|$ (D) If $\hat{\mathbf{u}}$ lies in the xz-plane then $2|\mathbf{u}_1| = |\mathbf{u}_3|$
- Sol. (B, C)

$$\hat{\mathbf{w}}.(\hat{\mathbf{u}}\times\vec{\mathbf{v}})=1$$

$$\Rightarrow |\hat{\mathbf{w}}||\hat{\mathbf{u}}\times\vec{\mathbf{v}}|\cos\alpha=1$$

$$\cos \alpha = 1$$

$$\Rightarrow \hat{\mathbf{w}} \perp \hat{\mathbf{u}}$$
 and $\hat{\mathbf{w}} \perp \vec{\mathbf{v}}$

as it is given there exist a vector \vec{v}

$$\Rightarrow$$
 \hat{w} must be \perp to \hat{u}

hance infinite many v exists

if
$$\hat{\mathbf{u}} = \mathbf{u}_1 \hat{\mathbf{i}} + \mathbf{u}_2 \hat{\mathbf{j}}$$

$$\overrightarrow{u} \cdot \overrightarrow{w} = 0 \implies (u_1 + u_2) = 0$$

$$\Rightarrow |u_1| = |u_2|$$

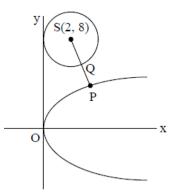
if
$$u = u_1 \hat{i} + u_2 \hat{k}$$

$$\vec{u} \cdot \vec{w} = 0$$

$$u_1 + 2u_3 = 0$$

$$\Rightarrow$$
 $|u_1| = 2|u_3|$.

47. Let P be the point on the parabola $y^2 = 4x$ which is at the shortest distance from the centre S of the circle $x^2 + y^2 - 4x - 16y + 64 = 0$. Let Q be the point on the circle dividing the line segment SP internally. Then



(A) SP =
$$2\sqrt{5}$$

(B)
$$SQ + QP = (\sqrt{5} + 1) : 2$$

(C) the x-intercept of the normal to the parabola at P is 6

(D) the slope of the tangent to the circle at Q is
$$\frac{1}{2}$$

Equation of normal of parabola is $y + tx = 2t + t^3$

Normal passes through S(2, 8)

$$\Rightarrow$$
 t = 2

Hence
$$P = (4, 4)$$
 and $SQ = radius = 2$

48. Let
$$a, b \in \square$$
 and $a^2 + b^2 \neq 0$. Suppose $S = \left\{ z \in \square : z = \frac{1}{a + ibt}, t \in \square, t \neq 0 \right\}$, where $i = \sqrt{-1}$.

If z = x + iy and $z \in S$, then (x, y) lies on

(A) the circle with radius
$$\frac{1}{2a}$$
 and centre $\left(\frac{1}{2a},0\right)$ for $a>0,b\neq0$

(B) the circle with radius
$$-\frac{1}{2a}$$
 and centre $\left(-\frac{1}{2a},0\right)$ for a < 0, b \neq 0

(C) the x-axis for
$$a \neq 0, b = 0$$

(D) the y-axis for
$$a = 0$$
, $b \neq 0$

$$x + iy = \frac{1}{a + ubt}$$

$$x^2 + y^2 - \frac{x}{a} = 0$$

(A) Centre
$$\left(\frac{1}{2a},0\right) r = \frac{1}{2a}a > 0$$

(B) Centre
$$\left(\frac{1}{2a}, 0\right) r = -\frac{1}{2a}a < 0$$

(C) x-axis
$$x = \frac{1}{a}, b = 0$$

(D) y-axis
$$y = \frac{1}{bt}, a = 0$$

49. Let
$$a, \lambda, \mu \in \square$$
. Consider the system of linear equations

$$ax + 2y = \lambda$$

$$3x - 2v = \mu$$

Which of the following statement(s) is (are) correct?

(A) If a =
$$-3$$
, then the system has infinitely many solutions for all values of λ and μ

(B) If
$$a \neq -3$$
, then the system has a unique solution for all values of λ and μ

(C) If
$$\lambda + \mu = 0$$
, then the system has infinitely many solutions for $a = -3$

(D) If $\lambda + \mu \neq 0$, then the system has no solution for a = -3

System has unique solution for $\frac{a}{3} \neq \frac{2}{-2}$

system has infinitely many solutions for $\frac{a}{3} = \frac{2}{-2} = \frac{\lambda}{\mu}$ and no solution for $\frac{a}{3} = \frac{2}{-2} \neq \frac{\lambda}{\mu}$

50. Let $f: \left[-\frac{1}{2}, 2\right] \to \Box$ and $g: \left[-\frac{1}{2}, 2\right] \to \Box$ be functions defined by $f(x) = \left[x^2 - 3\right]$ and

g(x) = |x| f(x) + |4x - 7| f(x), where [y] denotes the greatest integer less than or equal to y for $y \in \Box$. Then

- (A) f is discontinuous exactly at three points in $\left[-\frac{1}{2},2\right]$
- (B) f is discontinuous exactly at four points in $\left[-\frac{1}{2},2\right]$
- (C) g is NOT differentiable exactly at four points in $\left(-\frac{1}{2},2\right)$
- (D) g is NOT differentiable exactly at five points in $\left(-\frac{1}{2},2\right)$

$$f(x) = [x^2 - 3]$$

Which is discontinuous at x = 1, $\sqrt{2}$, $\sqrt{3}$, 2

$$g(x) = f(x) [|x|+|4x-7|]$$

f(x) is non differentiable at $x = 1, \sqrt{2}, \sqrt{3}$

& |x| + |4x - 7| is non differentiable at $x = 0, \frac{7}{4}$ success Educate PVL LE

But
$$f(x) = 0 \forall x \in [\sqrt{3}, 2)$$

Hence g(x) is non differentiable x = 0, 1 $\sqrt{2}$, $\sqrt{3}$.

Integer type

PARAGRAPH 1

Football teams T_1 and T_2 have to play two games against each other. It is assumed that the outcomes of the two games are independent. The probabilities of T_1 winning, drawing and losing a game against T_2 are $\frac{1}{2}$, $\frac{1}{6}$ and $\frac{1}{3}$, respectively. Each team gets 3 points for a win, 1 point for a draw and 0 point for a loss in a game. Let X and Y denote the total points scored by teams T_1 and T_2 , respectively, after two games.

51. P(X > Y) is

(A)
$$\frac{1}{4}$$

(B)
$$\frac{5}{12}$$

(C)
$$\frac{1}{2}$$

(D)
$$\frac{7}{12}$$

Sol. (B)

 $P(X > Y) = P(T_1 \text{ wins both}) + P(T_1 \text{ wins either of the matches and other is draw})$

$$=\frac{1}{2} \times \frac{1}{2} + 2 \times \frac{1}{2} \times \frac{1}{6} = \frac{1}{4} + \frac{1}{6} = \frac{5}{12}$$

52. P(X = Y) is

(A)
$$\frac{11}{36}$$

(B)
$$\frac{1}{3}$$

(C)
$$\frac{13}{36}$$

(D)
$$\frac{1}{2}$$

Sol. (C)

 $P(X = Y) = P(T_1 \text{ and } T_2 \text{ win alternately}) + P(Both matches are draws)$

$$=2\times\frac{1}{2}\times\frac{1}{3}+\frac{1}{6}\times\frac{1}{6}=\frac{1}{3}+\frac{1}{36}=\frac{13}{36}$$

PARAGRAPH 2

Let $F_1(x_1,0)$ and $F_2(x_2,0)$, for $x_1<0$ and $x_2>0$, be the foci of the ellipse $\frac{x^2}{\alpha}+\frac{y^2}{\alpha}=1$. Suppose a parabola having vertex at the origin and focus at F2 intersects the ellipse at point M in the first quadrant and at point N in the fourth quadrant.

53. The orthocenter of the triangle F₁MN is

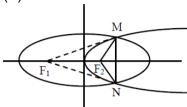
$$(A)\left(-\frac{9}{10},0\right)$$

(B)
$$\left(\frac{2}{3},0\right)$$

(C)
$$\left(\frac{9}{10},0\right)$$

(C)
$$\left(\frac{9}{10}, 0\right)$$
 (D) $\left(\frac{2}{3}, \sqrt{6}\right)$

(A) Sol.



$$e = \frac{1}{3}$$

Parabola :
$$y^2 = 4x$$

M and N are
$$\left(\frac{3}{2}, \sqrt{6}\right)$$
 & $\left(\frac{3}{2}, -\sqrt{6}\right)$

For orthocentre: one altitude is y = 0 (MN is perpendicular)

other altitude is :
$$(y - \sqrt{6}) = \frac{5}{2\sqrt{6}} \left(x - \frac{3}{2}\right)$$

orthocentre is
$$\left(-\frac{9}{10},0\right)$$

54. If the tangents to the ellipse at M and N meet at R and the normal to the parabola at M meets the xaxis at Q, then the ratio of area of the triangle MQR to area of the quadrilateral MF₁NF₂ is

Sol.

Equation of tangent at M and N are $\frac{x}{6} \pm \frac{y\sqrt{6}}{2} = 1$

R(6, 0)

Equation of normal $(y - \sqrt{6}) = -\frac{\sqrt{6}}{2} \left(x - \frac{3}{2} \right)$

$$Q\left(\frac{7}{2},0\right)$$

Area of
$$\triangle MQR = \frac{1}{2} \times \sqrt{6} \times \frac{5}{2} = \frac{5\sqrt{6}}{4}$$

Area of
$$MF_1NF_2 = \frac{\sqrt{6}}{2} + \frac{3\sqrt{6}}{2} = \frac{4\sqrt{6}}{2}$$

Ratio:
$$\frac{5\sqrt{6}}{4}$$
: $\frac{4\sqrt{6}}{2} = \frac{5}{8}$.